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NEWS 21 MAR 23 CA/CAPLUS enhanced with more than 250,000 patent
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enhanced
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NEWS 25 APR 24 CA/CAPLUS now has more comprehensive patent assignee

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 assignment/reassignment information
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 NEWS 29 APR 28 Limits doubled for structure searching in CAS
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NEWS EXPRESS JUNE 27 08 CURRENT WINDOWS VERSION IS V8.3,
 AND CURRENT DISCOVER FILE IS DATED 23 JUNE 2008.

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FILE COVERS 1907 - 28 Apr 2009 VOL 150 ISS 18
 FILE LAST UPDATED: 27 Apr 2009 (20090427/ED)

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 reclassification data for the third quarter of 2008.

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This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s prepar? (bisphenol (w) a)
MISSING OPERATOR 'PREPAR? (BISPENOL'
The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

```
=> s prepar? (s) (bisphenol (w) a)
    148790 PREPAR?
    147140 PREP
    2499 PREPS
    149408 PREP
        (PREP OR PREPS)
2266330 PREPD
    3 PREPDS
2266332 PREPD
        (PREPD OR PREPDS)
2459863 PREPAR?
        (PREPAR? OR PREP OR PREPD)
    81242 BISPENOL
    5064 BISPENOLS
    82769 BISPENOL
        (BISPENOL OR BISPENOLS)
23280567 A
L1      8851 PREPAR? (S) (BISPENOL (W) A)

=> s l1 (L) rectification
    19023 RECTIFICATION
    116 RECTIFICATIONS
    19090 RECTIFICATION
        (RECTIFICATION OR RECTIFICATIONS)
L2      2 L1 (L) RECTIFICATION
```

=> d l2 1-2 ibib abs

```
L2  ANSWER 1 OF 2  CAPLUS  COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:  2005:41403  CAPLUS
DOCUMENT NUMBER:   142:375820
TITLE:             New sideline extraction process for catalytic
                   rectification
INVENTOR(S):       Qiu, Zhaorong; Wang, Cheli; Cheng, Minlian; Ye, Qing;
                   Yang, Jihe
PATENT ASSIGNEE(S): China Petrochemical Co., Ltd., Peop. Rep. China;
                   Jiangsu Petrochemical College
SOURCE:            Faming Zhuanli Shenqing Gongkai Shuomingshu, 25 pp.
                   CODEN: CNXXEV
DOCUMENT TYPE:     Patent
LANGUAGE:          Chinese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:
```

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|------------|------|----------|-----------------|----------|
| ----- | ---- | ----- | ----- | ----- |
| CN 1478577 | A | 20040303 | CN 2002-142233 | 20020827 |

CN 1247289 C 20060329 CN 2002-142233 20020827
 PRIORITY APPLN. INFO.:
 AB The sideline extraction method for drawing the product and/or byproduct out during catalytic rectification by mounting an extractor mounted on the middle of the reaction region of the catalytic rectification tower is presented. The systems used include a solid-liquid system, a liquid-liquid system or its layered alternative, or a liquid-gas system. The liquid in the solid-liquid system may be separated by gravity separation method or filtration and fed back to the reaction region. The liquid-liquid system may be separated by membrane filtration, rectification, extraction, adsorption, absorption, gas stripping, etc., and one kind of liquid in the liquid-liquid system may be fed back to the reaction region, while the layered liquid-liquid system may be separated by gravity separation. The extractor for the liquid-liquid system is an internal liquid separator and an external liquid separator. An internal cooling separator is mounted in the top of the catalytic rectification tower, and used to cool and sep. the gas phase in the rectification tower. The method may be used in esterification, transesterification, saponification, hydrolysis, alkylation, isomerization, amination, oxidation, etherification, etc. Tri-Bu citrate, isobutylene, and bisphenol A were prepared by using the sideline extraction process.

L2 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1988:168545 CAPLUS
 DOCUMENT NUMBER: 108:168545
 ORIGINAL REFERENCE NO.: 108:27719a, 27722a
 TITLE: Process for producing polycarbonates which do not cause corrosion during molding
 INVENTOR(S): Koga, Shinichiro; Matsuno, Akira; Sakata, Katsuyuki; Otani, Yoshiaki; Akihara, Isao
 PATENT ASSIGNEE(S): Mitsubishi Chemical Industries Co., Ltd., Japan
 SOURCE: Eur. Pat. Appl., 11 pp.
 CODEN: EPXXDW
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|---|------|----------|-----------------|------------|
| EP 251586 | A2 | 19880107 | EP 1987-305423 | 19870618 |
| EP 251586 | A3 | 19890125 | | |
| EP 251586 | B1 | 19920429 | | |
| R: DE, IT, NL | | | | |
| JP 62297320 | A | 19871224 | JP 1986-142164 | 19860618 |
| JP 06076482 | B | 19940928 | | |
| JP 62297321 | A | 19871224 | JP 1986-142165 | 19860618 |
| JP 06076483 | B | 19940928 | | |
| JP 63090536 | A | 19880421 | JP 1986-235829 | 19861003 |
| JP 03020132 | B | 19910318 | | |
| BR 8703052 | A | 19880308 | BR 1987-3052 | 19870617 |
| US 4839458 | A | 19890613 | US 1987-63001 | 19870617 |
| PRIORITY APPLN. INFO.: | | | | |
| | | | JP 1986-142164 | A 19860618 |
| | | | JP 1986-142165 | A 19860618 |
| | | | JP 1986-235829 | A 19861003 |
| AB Polycarbonates, which have reduced CC14 (I), which, upon molding, do not | | | | |

yellow or cause mold die corrosion are prepared by polymerizing a hydroxydiaryl compound, COC12, and, optionally, a diamine compound or an acid chloride in a CH2Cl2 solvent, where the concentration of I present in the COC12 and CH2Cl2 satisfies the equation $A + 5B < 200$ [$A = I$ concentration (ppm) in COC12; $B = I$ concentration (ppm) in CH2Cl2]. COC12 (containing 500 ppm I) was purified by rectification in a Widmer rectification pipe at column temperature 8° and reflux ratio 1:1, producing COC12 containing 5 ppm I. A polycarbonate was prepared by treating a solution containing 15.3 L H2O, 1.03 kg NaOH, 3.0 g Na hydrosulfite, and 2790 g bisphenol A, into which 3.1 L of an aqueous 10 wt/volume % NaOH solution and 7.65 L CH2Cl2 (containing 5 ppm I) were added, and 1.38 kg purified COC12 were blown in under stirring for 60 min, followed by addition of 66.7 g 4-tert-BuC6H4OH and 3.1 g Et3N. The mixture was stirred for 1 h during which polymerization

took

place. The organic layer was washed with H2O, neutralized with 3.0 L 10% H3PO4, washed with H2O three times, and heated to distill off CH2Cl2, producing a powdery polycarbonate which, after drying and pelletization had yellowness index 6. After molding 200 sheets from the pellets (using carbon steel molding dies; 300°) no molding die corrosion was observed

=> d his

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FILE 'CAPLUS' ENTERED AT 15:20:03 ON 28 APR 2009

L1 8851 S PREPARE? (S) (BISPHENOL (W) A)
L2 2 S L1 (L) RECTIFICATION

=> s l1 (l) (recycle (s) effluent)

33193 RECYCLE
2160 RECYCLES
34963 RECYCLE
(RECYCLE OR RECYCLES)
98794 EFFLUENT
51189 EFFLUENTS
132938 EFFLUENT
(EFFLUENT OR EFFLUENTS)

L3 0 L1 (L) (RECYCLE (S) EFFLUENT)

=> s l1 and (recycle (s) byproduct)

33193 RECYCLE
2160 RECYCLES
34963 RECYCLE
(RECYCLE OR RECYCLES)
40520 BYPRODUCT
33749 BYPRODUCTS
67357 BYPRODUCT
(BYPRODUCT OR BYPRODUCTS)

L4 339 RECYCLE (S) BYPRODUCT
1 L1 AND (RECYCLE (S) BYPRODUCT)

=> d l4 ibib abs

L4 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER: 2001:355038 CAPLUS
DOCUMENT NUMBER: 134:340822

TITLE: Preparation and crystallization process for the manufacture of high-purity bisphenol A
 INVENTOR(S): Heydenreich, Frieder; Prein, Michael; Boediger, Michael; Neumann, Rainer
 PATENT ASSIGNEE(S): Bayer A.-G., Germany
 SOURCE: Ger. Offen., 6 pp.
 CODEN: GWXXBX
 DOCUMENT TYPE: Patent
 LANGUAGE: German
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

| PATENT NO. | KIND | DATE | APPLICATION NO. | DATE |
|---|------|----------|------------------|----------|
| DE 19954786 | A1 | 20010517 | DE 1999-19954786 | 19991115 |
| WO 2001036358 | A1 | 20010525 | WO 2000-EP10827 | 20001103 |
| W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG AU 2001010277 A 20010530 AU 2001-10277 20001103 BR 2000015555 A 20020709 BR 2000-15555 20001103 EP 1232134 A1 20020821 EP 2000-971412 20001103 EP 1232134 B1 20040922 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR JP 2003523950 T 20030812 JP 2001-538314 20001103 AT 276990 T 20041015 AT 2000-971412 20001103 CN 1189438 C 20050216 CN 2000-815651 20001103 ES 2228621 T3 20050416 ES 2000-971412 20001103 TW 226323 B 20050111 TW 2000-89123660 20001109 IN 2002MN00520 A 20060505 IN 2002-MN520 20020422 MX 2002004812 A 20030128 MX 2002-4812 20020514 US 6710211 B1 20040323 US 2002-129944 20020826 PRIORITY APPLN. INFO.: DE 1999-19954786 A 19991115 WO 2000-EP10827 W 20001103 | | | | |

AB Highly pure bisphenol A, prepared by the condensation of phenol with acetone in the presence of an acidic sulfonated polystyrene resin cation exchanger catalyst, is purified by:
 (A) a primary crystallization in the form of a continuous or discontinuous layer
 (B) subjecting it to an optional distillation or crystallization; and
 (C) removing water, acetone, and phenol from the byproduct stream for recycle to the initial reactor. A process flow diagram is presented.

=> d his

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Serial No.: 10/573697

FILE 'CAPLUS' ENTERED AT 15:20:03 ON 28 APR 2009
L1 8851 S PREPARE? (S) (BISPHENOL (W) A)
L2 2 S L1 (L) RECTIFICATION
L3 0 S L1 (L) (RECYCLE (S) EFFLUENT)
L4 1 S L1 AND (RECYCLE (S) BYPRODUCT)

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